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CHARACTERIZATION OF BACTERIA FOUND IN METAL-WORKING FLUIDS  
AND THE WASTE TREATMENT SYSTEM INVOLVED IN DEGRADATION OF  
WASTE WATER

by

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THESIS

HQ. MILPERCEN

Submitted to the Graduate School  
of Wayne State University

Detroit, Michigan

in partial fulfillment of the requirements

for the degree of

MASTERS OF SCIENCE

1991

20030207002

MAJOR: BIOLOGICAL SCIENCES  
(Microbiology)

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*Adel Whinnell* 2/2/91  
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### DEDICATION

To my parents, Robert and Frances Cornwell. I would like to take this opportunity to thank you for everything that you have done for me.

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#### ACKNOWLEDGMENT

I wish to express my gratitude to my advisor, Dr. Harold W. Rossmore, whose continuous support has helped me accomplish my goal. Dr. Rossmore is a very knowledgeable and caring man. I would like to thank him for all his help.

I would also like to thank the U.S. Army for selecting me to attend graduate school under the fully funded advanced civilian schooling program. Without this selection I would not have attended graduate school when I did. In addition, I would like to thank the members of my committee, Dr. J. M. Jay and Dr. K. C. Chen.

I would like to take this opportunity to extend a very special thanks to Helen Douglas and Veronica Riha for their friendship, technical assistance and encouragement during my research period.

## FOREWORD

My main purpose for undertaking this study was to learn all I could in microbiology that would be beneficial to the U.S. Army. I was sent to Graduate School by the U.S. Army to attain a Masters in the field of microbiology. During my first semester at Wayne State University there were not many courses offered in microbiology and the only way I could think of to learn a considerable amount about microbiology was by undertaking a research project. At that time the Army was also interested in the degradation of oils and as a result of this I started this project; looking at the biodeterioration and biodegradation of metal working fluids. This project has helped me fulfill the requirements that I set upon myself; that of learning all I could in the field of microbiology with respect to biodeterioration and biodegradation of metal working fluids. This study and other laboratory work has introduced me to different techniques, equipment, and media that I would not have been exposed to if I had opted to undertake the Plan C Masters.

This thesis is written primarily for two audiences: the U.S. Army and Wayne State University. Thus, it must be remembered that persons not intimately involved in this field have a vested interest in my work.

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## INTRODUCTION

The work for this Masters Thesis involved the study of microorganisms in Metal Working Fluids (MWF) and the fluidized bed reactors involved in their disposal. It was of interest in determining what bacteria survive in MWF under different treatments, both chemical and biological, and during the waste treatment process. These studies focused on bacteria isolated from MWF and from different locations in the waste treatment system. Also looked was the survival of bacteria, under laboratory conditions, after changes in their environment. The environmental changes included: changes in pH, biocide concentration (biocides used were the same ones used in the industrial plant), and carbon source. All samples used for this study were collected from a General Motors plant (Delco Moraine New Departure Hyatt (NDH) plant) in Sandusky, Ohio.

Because of the possibility of future assignment in this field (biodegradation and/or degradation of oils) in the Army, an understand of what was occurring in these environments was needed by the author. From this research with MWF, results from these changes in the environment could be of interest to the Army. This study also helped one to understand the effects that changes in the environment had on the survival of bacteria. The areas of particular importance to this study were metal working fluids, biocides and the fluidized bed reactors used.

### Metal Working Fluids

Metal working fluids are used in many industries today. Some types of metal working or processing methods that use MWF are [3]:

- 1.) machining and grinding
- 2.) stamping, blanking, drawing, and spinning
- 3.) molding
- 4.) rolling.

The primary function of the MWF is to increase the efficiency of these and other industrial operations. The MWF is employed for different functions depending upon the type of machine operation to which it is applied. The two basic functions which any MWF must perform are [24]:

- 1.) To remove heat from the cutting tool, work piece and chips (i.e. to cool).
- 2.) To reduce the friction between the flowing chips and the cutting tool, thus reducing the heat generated in cutting.

The major benefit from these functions is to improve tool life. There are also other functions that the MWF must perform to increase the efficiency of operation. These include [5]:

- 1.) provide rust/corrosion protection
- 2.) lubricate exposed machine tool parts
- 3.) reduce distortion through cooling
- 4.) allow for increased tool speeds
- 5.) wash away chips from the work piece/tool interface.

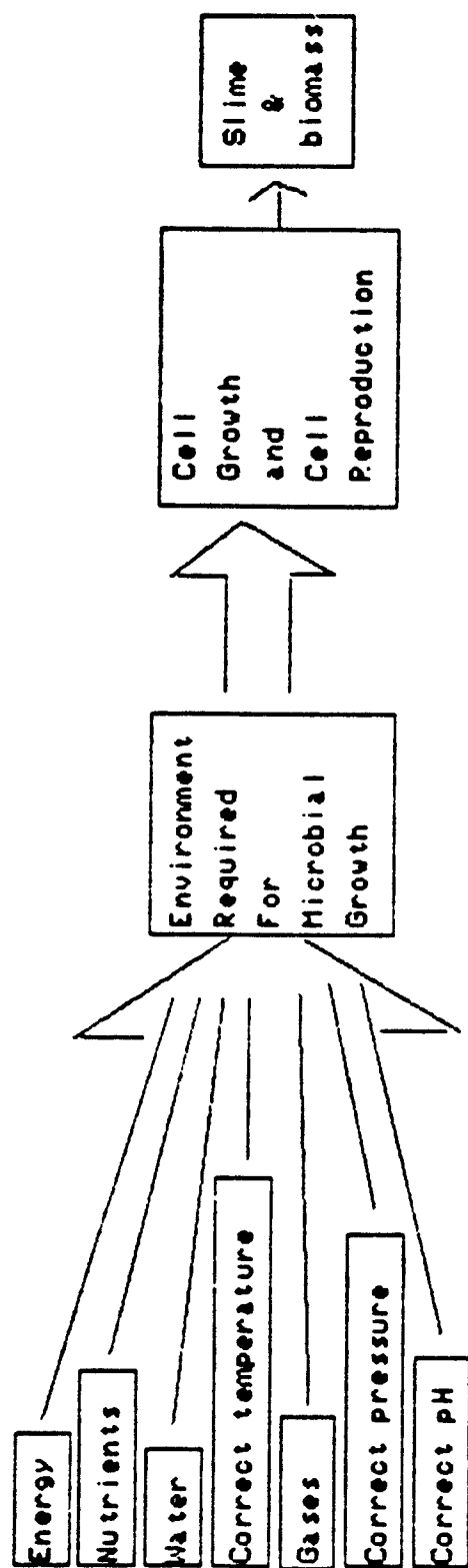
The type of MWF examined in this study was a water based MWF. With the addition of water to metalworking operations it has created a more than favorable environment for a variety of microorganisms. Since MWF is such an excellent environment for microbial growth, industries working with MWF are concerned about contamination. The massive amount of growth of microorganisms in MWF can cause adverse effects to the industrial operations. Some of the common problems due to microbial contamination are listed in Table 1 [20].

TABLE 1: PROBLEMS CAUSED BY MICROBIAL  
CONTAMINATION

- odor development
- decrease in pH
- changes in emulsion
- increase in corrosion rates
- changes in coolant chemistry
- decreased tool life
- surface-finish blemishes
- clogged filters, screens & lines
- increase work piece rejection rates

For growth of microorganisms in any environment, certain physiological and nutritional requirements must be met. Figure 1, shows the requirements for microbial growth [7]. If any one of these factors is lacking then cell growth stops. MWF becomes contaminated with microorganisms because the fluid contains all the required nutrients needed to support microbial growth. Some of the nutrients found in MWF are listed in Table 2 [20]. The mineral oil base

FIGURE 1: Physical and Nutritional Requirements for the Growth of Microorganisms in MWF.



stocks, glycols, fatty acid soaps, amines, and other constituents of MWF provide all of the essential nutrients required for microbial growth. The other requirements for microbial growth: pH, temperature, pressure, and gases, are met through the normal operation and maintenance of the MWF.

**TABLE 2: NUTRIENTS IN METAL WORKING FLUIDS**

<u>Organic</u>	<u>Inorganic</u>
	cation:
- mineral waxes	- iron
- fatty oil	- calcium
- fatty acid soaps	- sodium
- synthetic esters	- magnesium
- phosphate esters	- manganese
- amines	
	anions:
	- sulfate
	- chloride
	- phosphate

Large populations of microorganisms are readily detected because of their effects on the fluid or system (see Table 1). Unfortunately, by the time these effects are noticed, it may already be too late to rescue the system and prevent further damage. In addition, since microbes are so small and cannot be seen by the naked eye, the biodeterioration problems that they cause are often not recognized until it is too late [8]. The growth of the bacteria in the MWF can be controlled by the addition of preservatives. The preservatives that are of interest to this study are biocides.

## Biocides

Biocides are added to MWF to destroy/inhibit microbial cells. Some of the ways that biocides react (mode of action) with the microorganisms are described below. Two of the biocides used in this study, A and B, are considered formaldehyde (FA) releasers. This was demonstrated by the extensive work of Rossmore et al [22, 23]. It was found that the release of FA from the biocide was essential for its mode of action on the microbial cells. When FA was released, it reacted with essential nucleophiles (SH groups and amines) in the biological environment and then reacted with the cells causing loss of viability. Biocide C also reacts in a similar manner with the nucleophiles.

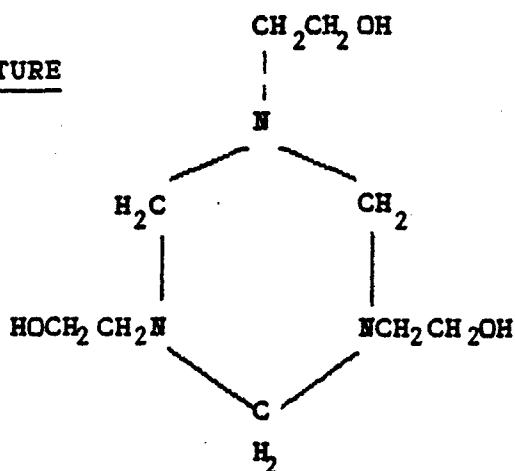
Another probable mode of action involves the biocide making contact with the cell envelope and thus gaining cell entry. Some entry can be contributed to the oil-water portion of the cell envelope, thus allowing the biocide to penetrate the lipophilic cell envelope. Some components of the biocides also have structures similar to the essential nutrients and metabolites; it is therefore possible to envision a form of competition for some of them [23]. As a result of the action of the biocide on the microbial cells, the cells may stop cell growth and reproduction.

The biocides in this study were used in the MWF in the General Motor Plant and these are outlined below.

1. Hexahydro-1,3,5-tris(2-hydroxyethyl)-s-triazine





STRUCTURE

**EFFICACY:** This biocide is an effective antimicrobial agent that can be used to inhibit the growth of bacteria in aqueous-based metal working fluids. It is effective against gram-positive and gram-negative bacteria. It was tested for efficacy at the recommended concentration of 1500 ppm (parts per million). The physical properties of this biocide are listed in Table 3 [13-15] and it is referred to as "Biocide A".

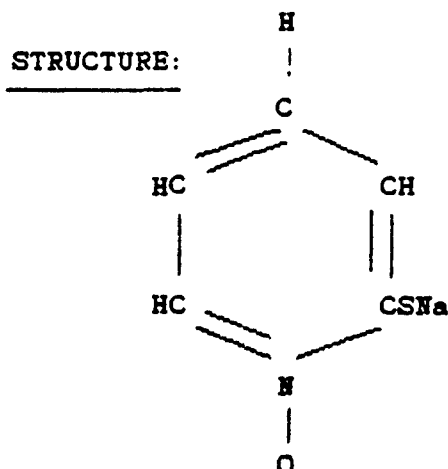
**TABLE 3: TYPICAL PHYSICAL PROPERTIES OF  
HEXAHYDRO-1,3,5-TRIS(2-HYDROXYETHYL)-  
S-TRIAZINE**

- active ingredient (%)	78.5
- inert ingredient (%)	21.5
- color	amber
- odor	faint amine

**2. 1-Hydroxy-2(1H)-Pyridinethion, Sodium salt and  
Hexahydro-1,3,5-tris(2-hydroxyethyl)-s-triazine**

This is a mixture of two other known biocides. The two components are:

a. Sodium 2-pyridinethiol-1-oxide ( $C_5H_4NOSNa$ )



b. Hexahydro-1,3,5-tris(2-hydroxyethyl)-s-triazine

( $C_9H_{21}N_3O_3$ ). See to Biocide A for structure.

**EFFICACY:** This biocide utilizes a potentially synergistic combination of an antifungal agent and an antibacterial agent. It provides pronounced growth inhibiting activity against gram-positive and gram-negative bacteria, yeasts, and molds. The antifungal agent is sodium-2-pyridinethiol-1-oxide and the antibacterial agent is hexahydro-1,3,5-tris(2-hydroxyethyl)-s-triazine. This biocide eliminates the need for adding two antimicrobial agents. It was tested for efficacy at the recommended concentration of 1000 ppm. Table 4 lists some of the typical properties of this biocide [16-18] and it is referred to as "Biocide B".

**TABLE 4: TYPICAL PHYSICAL PROPERTIES  
OF "BIOCIDE B"**

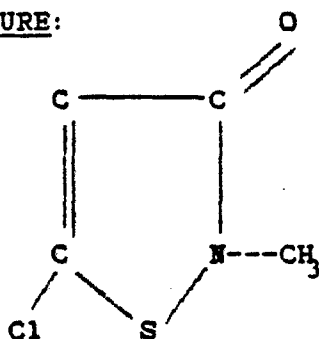
Active Ingredient (%)	70
Color	amber

3. This biocide is a mixture of 5-chloro-2-methyl-4-isothiazolin-3-one and 2-methyl-4-isothiazolin-3-one

a. 5-chloro-2-methyl-4-isothiazolin-3-one

(C<sub>4</sub>H<sub>4</sub>NOCl)

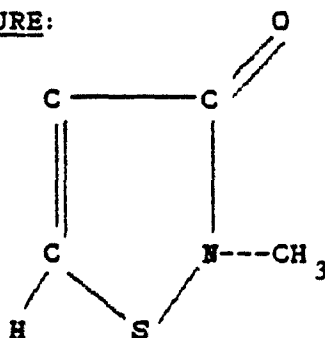
STRUCTURE:



b. 2-methyl-4-isothiazolin-3-one

(C<sub>4</sub>H<sub>5</sub>NO)

STRUCTURE:



EFFICACY: This biocide is used to inhibit the growth of bacteria and fungal contaminants in aqueous dilutions of emulsifiable synthetic and semi-synthetic MWF. Some of the

physical and chemical properties of this biocide are listed in Table 5 [21]. The recommended level of use as established for EPA regulation is 100 ppm. This biocide is referred to as "Biocide C".

**TABLE 5: PHYSICAL AND CHEMICAL PROPERTIES  
OF "BIOCIDE C"**

<b>Active Ingredients</b>	
- 5-chloro-2-methyl	
-4-isothiazolin-3-one	8.6 % min.
- 2-methyl-4-isothiazolin	
-3-one	2.6 % min.
<b>Appearance</b>	pale yellow
<b>Odor</b>	mild, aromatic

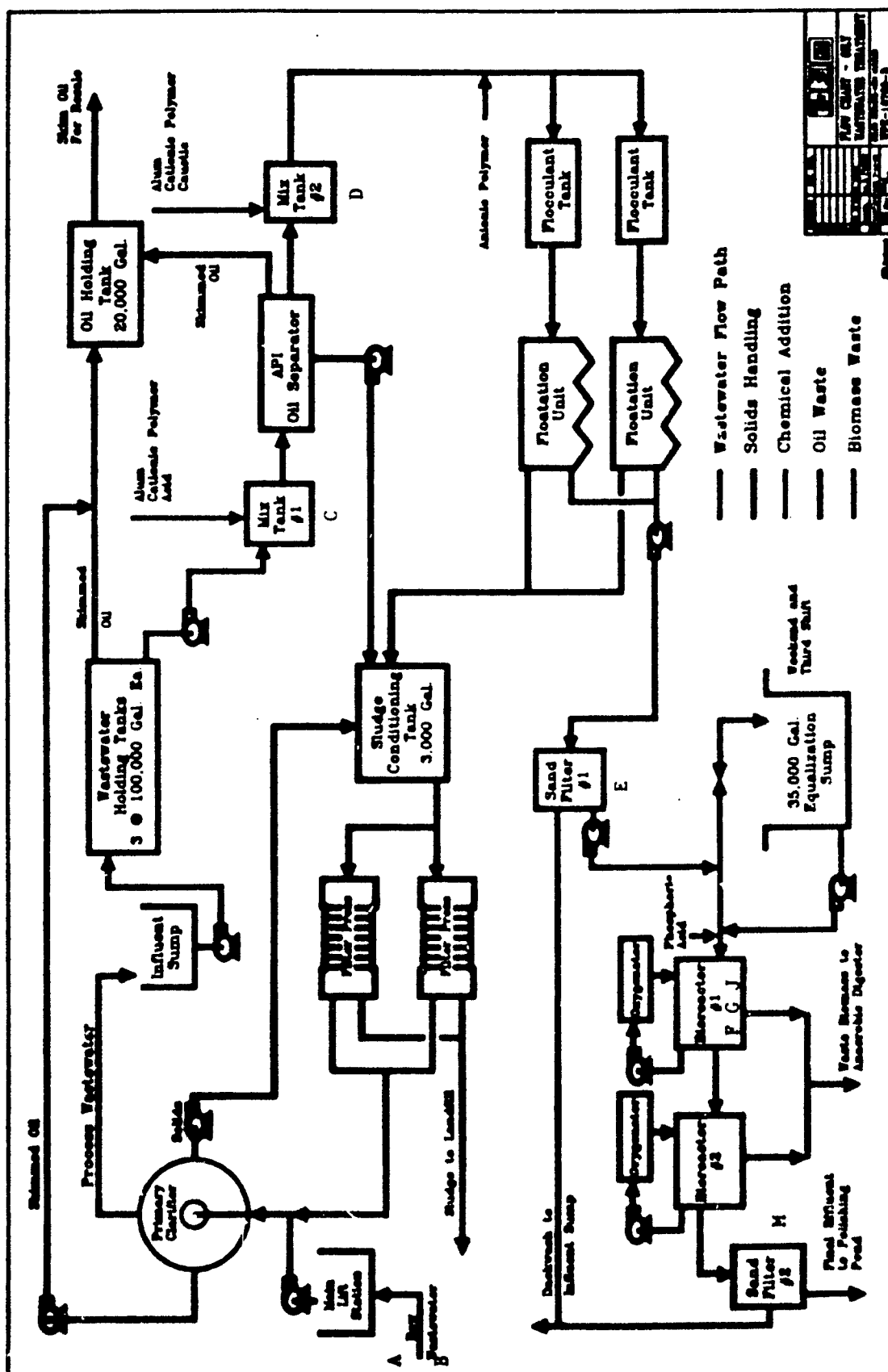
Because of the differences in location of their isolation, it was of interest to see if any of the microorganisms showed a difference in their minimal inhibitory concentration (MIC). The MIC is the concentration of an antimicrobial agent necessary to inhibit the growth of a particular strain of microorganism [1]. This test determines the concentration of an antimicrobial agent that is effective in preventing growth of the organisms and gives some indication of the minimal dosage that should be effective in controlling the microbes. It is important to realize that in a given species biological sensitivity can vary in its MIC by a factor of 2-5, i.e. the differences in their intrinsic resistance [9]. Many authors have reported the resistance of microorganisms to toxic chemicals and the ways in which microorganisms acquire their

resistance [6, 9, 10, 25, 26, 27]. It was desirable to see if the bacteria showed any differences in their MIC based on the location of the isolate. In addition, it was of interest to determine if these MIC differences were also evident in a medium stressing hydrocarbon utilization.

#### Waste Treatment System

The biological waste system used to biodegrade waste MWF was the OXITRON fluidized bed reactor. Waste MWF first passed through a physical-chemical treatment before it reached the fluidized bed reactor. Figure 2, shows a detailed schematic of the waste water process at Delco Moraine NDH Plant. This system is divided into a physical-chemical and a biological section.

The physical-chemical section involves seven steps from the primary clarifier to sand filter # 1. The primary clarifier is responsible for the removal of free oil and any solid material from the waste. The waste then goes into one of three holding tanks where it is allowed to sit for a period of time, allowing settling of more oil and its removal by skimming from the top. (The waste sits in one of the holding tanks while the other two are being filled and emptied of waste MWF.) After passing the holding tank, the waste water goes through a series of chemical treatments that further break down the waste material. In mix tank # 1, chemicals are added to lower the pH to approximately 4.8. This is done to split emulsion and further break down the



**Figure 2: Schematics of Waste Treatment System**

chemical components in the waste MWF. The waste then goes through another clarifier, the API oil separator, where more free oil is removed. Next is mix tank # 2, where more chemicals are added and this time the effect is to raise the pH to approximately 9.2. Finally, the waste material goes through the flotation units to remove any more solids that are brought to the top by dissolved air. From here the waste water material enters the biological portion of the system.

The Delco Moraine NDH system, where samples were collected, was one of the first General Motors Plant to install a full scale aerobic fluidized bed OXITRON system [12]. The OXITRON is a commercial embodiment of the aerobic biological fluidized bed process configuration system developed by Dori-Oliver, Inc. Figure 3 shows the OXITRON aerobic fluidized bed process schematics [12]. In this system, the waste water is passed upward through a rectangular or circular reactor containing a bed of sand (or granular activated carbon media may be used) at a velocity sufficient to expand the bed, resulting in a fluidized state. In this system, the combination of liquid and solid particles display what Leva [11] called particulate fluidization. The bed expands smoothly with none of the violent bubbling and particle motion characteristic of gas/solid fluidization, known as aggregative fluidization [11]. Once fluidized, the media particles provide a vast surface area for biological growth, leading to the

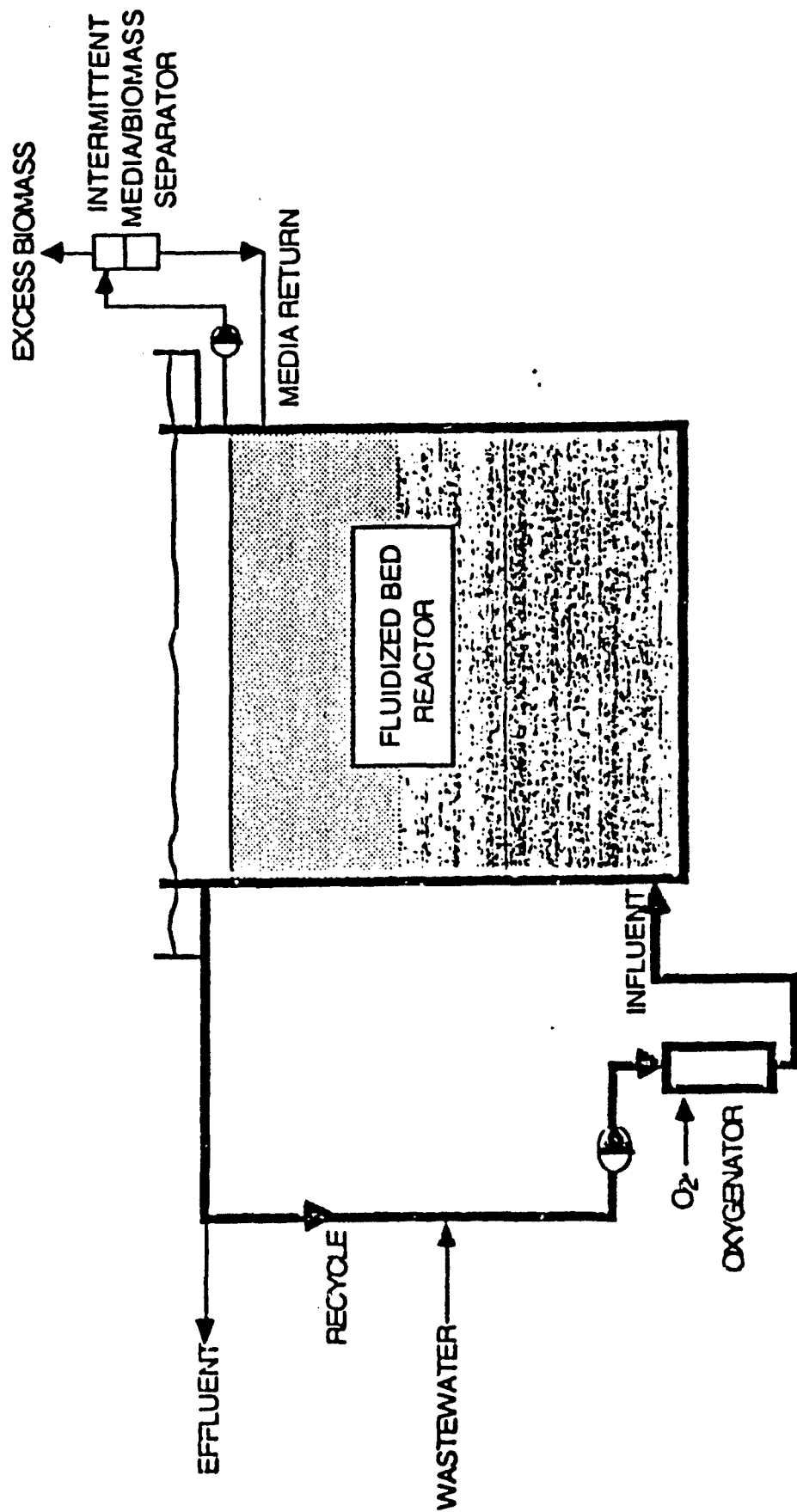


Figure 3: OXITRON Aerobic Fluidized Bed Process Schematic  
From Biological Fluidized Beds for Water and Waste Water Treatment



development of a biomass concentration approximately five to ten times greater than that normally maintained in conventional bioreactors. As a result of the biomass concentration on the media particles, they tend to become stratified in the bioreactor. It was shown that vertical density gradient could be determined as a result of the stratification [11]. The particles at the top of the fluidized bed had a greater biomass concentration than those at the bottom. Degradation of the waste water is accomplished by the microorganisms found in the biomass. Microbial degradation of complex organic compounds requires the maintenance of a long solid retention time in the biotreatment reactor, corresponding to a slow net growth rate of the microbial population. The high reactor biomass concentration allows achievement of a long solid retention time at a short liquid hydraulic retention time [28]. The use of dissolved oxygen in the influent stream satisfies the oxygen requirements.

The Delco Moraine NDH system was installed in late 1985 and early 1986 and consists of two fluidized bed reactors. The combine physical-chemical and biological process is designed to treat up to  $544 \text{ m}^3$ /day of industrial waste water [28]. The first reactor in this system is for carbonaceous oxidation and the second reactor is for nitrification. The effluent from the oily waste water treatment system is deficient in phosphorus; therefore, phosphorus is added to the first reactor in the form of phosphoric acid. In

addition, sodium hydroxide is added to buffer the carbon dioxide produced during the carbonaceous oxidation and also to buffer the hydrogen ions produced during the nitrification process.

The samples collected from the waste water treatment system are listed in Table 6. Refer to Figure 2 for location on the schematics chart. The samples sites were chosen based on the differences in the chemical treatment in the system.

TABLE 6: LOCATION OF THE SAMPLES COLLECTED

<u>Identifier</u>	<u>Location</u>
A	NVF from the north side
B	NVF from south side
C	Mix tank # 1
D	Mix tank # 2
E	Influent into sand filter # 1
F	Reactor 1 - free flowing liquid
G	Reactor 1 sand at 10 1/2 ft
H	10 1/2 ft sand gently shaken
I	10 1/2 ft sand sonic oscillation
J	Reactor 1 sand at 20 ft
K	20 ft sand gently shaken
L	20 ft sand sonic oscillation
M	Effluent - going to pond

#### Objectives

The main objective of this project was to characterize bacteria found in various locations of the waste water treatment and to determine whether organisms of the same species showed differences in response to the changes in their environment based on the location from which they were

isolated.

## METHODS AND MATERIALS

All tests performed in this research were run two or more times to reinforce the validity of findings.

### Microbiological Evaluation

Samples were collected from the General Motors Plant in plastic screwcap jars on 9/28/90 and 10/1/90. The samples were put on ice and transported to Wayne State University where microbiological evaluation was carried out. The bacterial count in the sample was determined by serial dilution on Plate Count Agar (PCA, Difco Laboratories, Detroit, MI) for all locations except location H, I, K, and L.

The samples from these location involved sand. Removal of bacteria from the sand was achieved by putting 1 gram of sand into a Zip-lock bag with 9 ml of sterile water. Removal was done by gently shaking the bag and another was also put in the sonic oscillator for 1 minute and subsequently diluted and plated. The plates were incubated at 33 C for 48 hours.

### Media and Culture Conditions

All cultures were maintained on tryptic soy agar (TSA, BBL, Detroit MI). The cultures were kept fresh by being transferred to new TSA every ten days. Twenty-four-hour-old cultures grown in tryptic soy broth (TSB, Difco Laboratory, Detroit, MI) were used for determining the pH range, MIC concentration and growth in the mineral salts based medium.

The bacterial population in the broth at 24 hour was approximately  $10^9$  bacteria/ml. Thus, the microbial population inoculum used in these tests was 0.1 ml of a 24-hour culture yielding a final level in the system of  $10^6$  CFU/ml.

#### Isolation & Identification of Organisms

Various colony types were cultured onto TSA. Colonies were gram stained for purity before being restreaked, to insure a pure culture stock. The bacterial isolates were all gram-negative; therefore, it was possible to use the PASCO I.D Tri-Panels for Gram-negatives to identify the microorganisms (see Appendix for instructions on the use of the PASCO panels).

#### Determining pH Range of Organisms

The pH sensitivity tests were done with both TSA and TSB. With the TSA, pH was adjusted to a range of 4-8 and with the TSB, the pH was adjusted to 3.5 to 9. Both media were sterilized before the pH was adjusted with HCl or NaOH. With the TSA, pH was tested with all the microorganisms to get the general pH range for the microorganism. The pH range was checked again, using TSB, to those microorganisms listed in table 17 to see if the results would be the same as those for TSA. It was of interest to see how acidic the media could be without affecting the survival of the bacteria.

Another test with pH variation was done on the sand from the bioreactor. Here, one wanted to see if the organisms still attached to the sand could survive a lower pH than those that were freed. The question asked of the author was "What would the pH have to be lowered to inhibit the growth of the bacteria still attached to the sand (biofilm)?"

To answer this question, a gram of sand from the bioreactor was placed into a 250 ml flask containing 100 ml sterile TSB. The pH of this medium then adjust to the following pH: 4.5, 5.0, 5.5, 6 and 7. The flasks were incubated at 33 C for 48 hours. Positive results or growth was indicated by turbidity of the medium.

#### Determining Biocide MIC

Based on the location of the bacteria isolated, it was also of interest to see the differences in MIC. These tests were performed using 0.1 ml of 24 hour old cultures grown in TSB and added to 10 ml of TSB. The tests were performed in standard test tubes set up for serial dilution. The amount of biocide added to the first test tube of TSB was 1/10 the published application concentration, which for biocide A was 1500 ppm (0.15 %) of biocide in solution, biocide B was 1000 ppm (0.1 %) of biocide in solution, and biocide C was 100 ppm (0.01 %) of biocide in solution. The dilutions used are listed in Table 7.

TABLE 7: Dilutions used in Biocide Testing (ppm)

Biocide A	100	50	25	12.5	6.25	3.125
Biocide B	100	50	25	12.5	6.25	3.125
Biocide C	10	5	2.5	1.25	.625	.3125

After these tests, a more definite end point for the MIC was determined. The dilutions used for these tests are listed in Table 8. For biocide A, a too low of a biocide concentration was used and the results were all positive.

TABLE 8: Dilutions of Biocides for End Point

Biocide A	10	7.5	5	3.75	2.5	1.875
Biocide B	100	75	50	37.5	25	18.75
Biocide C	10	7.5	5	3.75	2.5	1.875

#### Determining Growth of Organisms in Mineral Salts base Medium

The final experiment performed was to determine which microorganisms would grow in a mineral base medium described by Palleroni and Doudoroff [19]. This medium consisted of M/30 NaK phosphate buffer pH 6.8,  $\text{NH}_4\text{Cl}$  (0.1%),  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$  (0.05%), ferric ammonium citrate (0.005%),  $\text{CaCl}_2$  (0.0005%) and a carbon source (0.1%). The carbon source used in this case was the hydrocarbon n-hexadecane. The media was prepared and sterilized after which the hydrocarbon was added. The microorganisms used in this test are listed in

Table 20. The test was done in 100 ml flasks with 50 ml media and a 0.1 ml of 24 hour old culture sample. These were then put into a G24 environmental incubator shaker (New Brunswick Scientific), set at 28 C and 200 RPM. These samples were looked at daily for evidence of turbidity indicating growth. Those with growth were used for another test to see if they would still grow in this medium with the addition of Biocide C. Biocide C was selected because of its lower effective dose. The amount of Biocide C added was 2.5 ml of 100 ppm stock solution thus making the final biocide concentration 5 ppm. This amount was used because most of the bacteria tested survived at this concentration. The rationale behind this is that it was known that the microorganisms would grow in each test separately but could they survive in the mineral base medium plus the hydrocarbon and the biocide.



## RESULTS

### Bacterial Level

The results for the CFU/ml based on location are presented in Table 9. The results for the CFU/ml were obtained by selecting and counting the plates with 30 to 300 colonies. The Quebec counter was used to make the count. The formula used for determining CFU/ml was:

$$\text{CFU} = \frac{(\text{average number of colonies/plate})}{\text{ml} \quad (\text{dilution plated})(\text{volume plated in ml})}$$

Locations A through D have two sets of results because in the first set, it was believed that the agar was poured too hot based on the low count values. As outlined in Table 9, there was a slight increase in the second set of results compared to the first set. The results for A through D were much lower than the rest of the results because of their location. Location A through D are for MWF and for those collected from the physical-chemical portion of the waste treatment system.

For location D, the crowded plate method was used to determine the count because there was no growth on any of the higher dilution plates. The probable reason being, that the agar was poured too hot. The crowded plate method was done by counting the colonies in five square areas. These counts were then averaged and multiplied by sixty-four (area of the petri dish). A third count was not performed because the sample was five days old and the count would not have been as accurate.

TABLE 9: Results for Bacterial Colony Forming Units per Location		
Location	CFU/ml	Average
A	3	
	1st 89 x 10	5
	3	1.0 x 10
	118 x 10	
	3	
	2nd 191 x 10	5
	3	1.6 x 10
	122 x 10	
B	4	
	1st 14 x 10	5
	4	3.2 x 10
	32 x 10	
	4	
	2nd 165 x 10	6
	4	1.7 x 10
	169 x 10	
C	2	
	1st 153 x 10	4
	2	1.3 x 10
	109 x 10	
	2	
	2nd 251 x 10	4
	2	2.6 x 10
	262 x 10	
D	2	
	1st 36 x 10	3
	2	3.8 x 10
	39 x 10	
	2	
	2nd 1433 x 10	5
	2	1.4 x 10
	1267 x 10	
		done by crowded plate method

TABLE 9 (cont.): Results for Bacterial Colony Forming  
Units per Location

Location	CFU/ml	Average
E	5	6 6.4 x 10
	65 x 10	
	5	
	63 x 10	
F	5	7 1.3 x 10
	108 x 10	
	5	
	159 x 10	
H CFU/gm sand	6	7 3.8 x 10
	51 x 10	
	5	
	243 x 10	
I CFU/gm sand	7	9 3.0 x 10
	296 x 10	
	TNTC	
K CFU/gm sand	5	7 1.3 x 10
	98 x 10	
	5	
	161 x 10	
L CFU/gm sand	7	8 7.3 x 10
	44 x 10	
	7	
	101 x 10	
M	3	5 1.5 x 10
	202 x 10	
	3	
	97 x 10	
G - Did not do because H & I are the results using different removal techniques.		
J - Did not do because K & L are the results using different removal techniques.		

The results between the gently shaking and sonic removal in H & I and K & L were what one expected. The removal by the sonic bath removed more bacteria from the sand than by just gently shaking the sample. The first thought that one would think about the results for location M was that the count was too low. This was the sample that was going to the pond and microorganisms found here were the free living bacteria from the bioreactors. These were the bacteria that Andrews described as being individual cells that were so small and light that they were washed out of the bioreactors [2]. The overall results were what one expected based on the locations that they were collected.

#### Identification of Microorganisms

The results for the identification by the PASCO ID TRI Panels are given in Table 10. The results are listed by biotype number first and then by genus and species name if the biotype number matched up in the PASCO I.D. book. From the results, it showed that it was not always possible to isolate specific microorganism in each area. However, Acinetobacter lwoffii was found in most of the locations and it was this microorganism that was tested.

There are two possibilities for the failure cited above. First, all the microorganisms possible from each location were not isolated. If one had been selective, there might have been greater success in the isolation of microorganisms common to each location. The second reason for not getting

TABLE 10: The Results of the Identification of Microorganisms by the PASCO ID TRI-Panels

Location								
Bacteria #	A	B	C	D	E	F	H	
1	000004010	000004401 Aci. lwoffii	075430302	000000100 Aci. lwoffii	000000000 Aci. lwoffii	000000000 Aci. lwoffii	200700200	
2	000004012 Pseud. spp.	000044004	265740613	000000401	000000000 Aci. lwoffii	000000020 Moraxella		
3	000040037	200746216	377700000	000004000 Aci. lwoffii	000000000 Aci. lwoffii	000000020 Moraxella		
4	240044012	240746026		000004012 Pseud. spp.	000000040 Aci. lwoffii	000000040 Aci. lwoffii		
5	352005207	241006000		200000020	000000120			
6	366005005	252001401		200004020	100000000			
7	366400400	377700401		200004020	360744000			
8	4 weren't ID	377700000	3 weren't ID			1 wasn't ID	2 weren't ID	
9		770007412						

NOTE: The names below the biotype number are only listed for the ones with a sure match in the ID book. I also listed the number of microorganisms that I could not identify because they would not grow in the positive control well of the PASCO ID Tri-panels.

TABLE 10: Continuation of Identification of Microorganisms						
Location						
Bacteria #	I	K	L	M		
1	000000002 Aci. Iwoffi	000000002 Aci. Iwoffi	000740020	000000000 Aci. Iwoffi		
2	000000020 Moraxella	040000620	040004000	000000000 Aci. Iwoffi		
3	000017005	240000042	240540620	000000000 Aci. Iwoffi		
4	000075441			710000002		
5	200300020					
6	377700002	1 wasn't ID	1 wasn't ID	1 wasn't ID		
7	1 wasn't ID					

See note on first page of table 10.

a common microorganisms in each location might be due to the fact that some microorganism were only present in low numbers in that specific location. As it was stated earlier, Acinetobacter lwoffii was found in all locations except in Location C (mix tank # 1) where the pH is lowered to 4.8 due to the addition of chemical. It is possible that this organisms might not of survived this treatment of Location C. From here on out, the various tests were done with mostly Acinetobacter lwoffii and a few other bacteria.

#### Biocide MIC Results

For the biocide testing ten isolates were selected that keyed to Acinetobacter lwoffii from different locations (two of the samples used looked similar to Acinetobacter lwoffii based on colony characteristics) and challenged them with the three biocides. Table 11 gives the ten bacteria used for these tests by biotype number and their location. The results for these ten bacteria are presented in Tables 12 through 16. The interpretation of each table is given below its respective table. The MIC for each biocide is much less than the recommended level for industrial application.

MIC tests with Biocides B and C on different microorganisms from various locations to see if their MIC was essentially the same as that for Acinetobacter lwoffii. Table 17 gives the bacteria used for these tests by biotype and location. Tables 18 and 19 give the results of these tests, with interpretations below each table.

TABLE 11: Bacteria used in the MIC Testing			
Bacteria number	Biotype number	Location	Bacteria name
1		H	resembled Aci. lwoffii
2	000000000	F	Aci. lwoffii
3	000000000	E	Aci. lwoffii
4	000000000	E	Aci. lwoffii
5	000000000	M	Aci. lwoffii
6	000000002	I	Aci. lwoffii
7	000000401	D	Aci. lwoffii
8	000004401	B	Aci. lwoffii
9		A	resembled Aci. lwoffii
10	000000000	F	Aci. lwoffii

Bacteria 1 & 9 did not grow in the positive control well of the PASCO ID Tri-panels.



TABLE 12: Results of MIC Testing with "Biocide A"

Bacteria ----- [ ] ppm	1	2	3	4	5	6	7	8	9	10
3.125	++	++	++	++	++	--	++	++	++	++
6.25	++	++	++	++	++	--	++	++	++	++
12.5	++	++	++	++	++	--	++	++	++	++
25.0	++	++	++	++	++	--	++	++	++	++
50.0	++	++	++	++	++	--	++	++	++	++
100.0	++	--	+-	+-	--	--	--	--	--	--

+ means positive for turbidity

- means negative for turbidity

INTERPRETATION: The MIC for these microorganisms was greater than 50.0 ppm, Bacteria # 6 did not grow in the lowest concentration and it was assumed that the beginning inoculation was not viable. For the endpoint MIC, the tests were run using too low of a concentration and the results were all positive so the results were not included. This biocide was the last one tested so not as many microorganisms were tested with it.

TABLE 13: Results of MIC Testing with "Biocide B"

Bacteria ----- [ ] ppm	1	2	3	4	5	6	7	8	9	10
3.125	++	++	++	++	++	++	++	++	++	++
6.25	++	++	++	++	++	++	++	++	++	++
12.5	++	++	++	++	++	++	++	++	++	++
25.0	++	++	++	++	++	--	--	++	++	++
50.0	++	--	++	++	--	--	--	--	++	--
100.0	--	--	--	--	--	--	--	--	--	--

+ means positive for turbidity

- means negative for turbidity

INTERPRETATION: The results from this test showed that the MIC for this biocide in most bacteria tested here was greater than 25 ppm. For the results of a finer MIC endpoint refer to table 14. This biocide was also tested with other bacteria from various locations and the results are given in table 18.

TABLE 14: Results of MIC Testing with "Biocide C"

Bacteria ----- [ ] ppm	1	2	3	4	5	6	7	8	9	10
0.3125	++	++	++	++	++	++	++	++	++	++
0.625	++	++	++	++	++	++	++	++	++	++
1.25	++	++	++	++	++	++	++	++	++	++
2.50	++	++	++	++	++	++	+-	++	++	++
5.0	++	++	++	++	+-	++	--	++	++	++
10.0	--	+-	+-	+-	--	--	--	--	--	--

+ means positive for turbidity

- means negative for turbidity

INTERPRETATION: See Table 15 for the a finer range for the MIC endpoint. In these results it shows that the MIC is greater than 5.0 ppm in most of the bacteria. This biocide was used on other bacteria from the different locations, the results for these bacteria are shown in Table 19.

TABLE 15: Results of MIC Endpoint with "Biocide B"										
Bacteria	1	2	3	4	5	6	7	8	9	10
[ ] ppm										
18.75	++	++	++	++	++	--	++	++	++	++
25.0	++	++	++	++	++	--	++	++	++	++
37.5	--	--	++	++	--	--	++	--	--	--
50.0	--	--	--	--	--	--	+-	--	--	--
75.0	--	--	--	--	--	--	--	--	--	--
100.0	--	--	--	--	--	--	--	--	--	--

+ means positive for turbidity

- means negative for turbidity

INTERPRETATION: Bacteria # 6 did not grow in the lowest concentration in this testing but it grow in the previous testing with this biocide. It is assumed that this bacteria was not viable when the tests was started. The final MIC endpoints for the rest of the bacteria was greater than 25.0 ppm. See the table for the exact endpoint for each bacteria.

TABLE 16: Results of MIC Endpoint with "Biocide C"

Bacteria ----- [ ] ppm	1	2	3	4	5	6	7	8	9	10
1.875	++	++	++	++	++	--	++	++	++	++
2.5	++	++	++	++	++	--	++	++	++	++
3.75	++	++	++	++	++	--	++	++	++	++
5.0	++	++	++	++	++	--	--	++	+-	++
7.5	+-	+-	++	++	+-	--	+-	--	--	+-
10.0	--	+-	+-	++	+-	--	--	--	--	+-

+ means positive for turbidity  
 - means negative for turbidity

INTERPRETATION: Bacteria # 6 did not show any growth in this test but it showed growth in the first testing with "Biocide C", so it is assumed that this bacteria was not viable to begin with. The rest of the bacteria show a final MIC endpoint greater than 5.0 ppm except bacteria # 7 and 9. See the results for accurate endpoint.

TABLE 17: Bacteria used in MIC Testing for Table 18 and 19			
Bacteria Number	Biotype Number	Location	Name
1	252001401	B	Aci. anitratus
2	240746026	B	
3	377700401	B	
4	200746216	B	
5	366400400	A	
6	240044012	A	
7	000040437	A	
8	352005207	A	
9	241006000	B	
10	000004010	A	Pseud. spp.
11	000004012	A	
12	no growth	A	
13	366005005	A	Pseud. spp.
14	000004012	D	
15	377700000	C	
16	no growth	A	Moraxella
17	000000120	E	
18	040000620	K	
19	000740020	L	
20	000000022	I	
21	040004000	L	

**NOTE:** The words "no growth" indicate that these bacteria did not grow in the control well of the PASCO system, therefore a biotype number could not be determined. The names of the microorganisms are only provided for the ones whose biotype number keyed to a name.

TABLE 18: Results of MIC Testing with "Biocide B"  
with other Isolated Microorganisms

[ ] ppm bacteria						
	100	50	2.5	12.5	6.25	3.125
1	--	++	++	++	++	++
2	--	++	++	++	++	++
3	--	++	++	++	++	++
4	--	++	++	++	++	++
5	++	++	++	++	++	++
6	--	++	++	++	++	++
7	--	--	++	++	++	++
8	--	--	++	++	++	++
9	--	--	+-	++	++	++
10	--	++	++	++	++	++
11	--	++	++	++	++	++
12	--	--	--	++	++	++
13	--	--	++	++	++	++
14	--	++	++	++	++	++
15	--	++	++	++	++	++
16	--	--	--	++	++	++
17	--	--	++	++	++	++
18	--	--	--	++	++	++
19	--	--	--	++	++	++
20	--	--	++	++	++	++
21	--	--	+-	++	++	++

INTERPRETATION: As the results indicated, most of the bacteria have a MIC greater than 25 ppm. There a'iso was a variance between the microorganisms that might be a result due to the location that the bacteria were isolated from. Compared to the Acinetobacter lwoffii, some of these bacteria are more resistant to this biocide.

TABLE 19: Results of MIC Testing with "Biocide C"  
with other Isolated Microorganisms

<u>[1 ppm</u> bacteria	10	5	2.5	1.25	.62	.31
1	--	++	++	++	++	++
2	--	--	++	++	++	++
3	--	--	--	++	++	++
4	--	--	++	++	++	++
5	--	--	--	++	++	++
6	--	--	++	++	++	++
7	--	--	++	++	++	++
8	--	--	++	++	++	++
9	--	--	--	++	++	++
10	--	++	++	++	++	++
11	--	--	+-	++	++	++
12	--	+-	++	++	++	++
13	--	--	++	++	++	++
14	++	++	++	++	++	++
15	--	++	++	++	++	++
16	--	--	--	--	--	+-
17	--	--	--	++	++	++
18	--	--	--	--	--	++
19	--	--	--	--	++	++
20	--	--	--	--	++	++
21	--	--	--	--	++	++

INTERPERTATION: As the results indicates there is a variance in the data based upon the bacteria being tested. The results from the previous tests were using the same microorganisms but from different locations. The results shown here are different microorganisms and different locations. Compared to *Acinetobacter lwoffii* results, the bacteria here are more sensitive to this biocide.



### PH Range Results

The pH tests were performed on all of the bacteria isolated on TSA. Bacteria listed in Table 17 were also tested in TSB to see if there was a differences based on the medium being a liquid or solid. The results from both tests showed that most of the bacteria survived a pH range of 6 to 8. Some of the Bacteria tested in TSB also grew at pH 9 but not any lower than pH 6.

The results of the pH tests done on the sand from the bioreactor showed that they could grow at a lower pH than the isolated bacteria. There was obvious growth, based on turbidity, in pH 7, 6, 5.5 and 5. There was survival in pH 4.5, determined by surface plating on TSA, not by obvious growth. The bacteria isolated from surface plating were also used in subsequent studies in mineral salts-hydrocarbon medium.

Results indicated that the organisms attached to the sand could survive a lower pH environment then the isolated microorganisms. The question asked "How much would you have to lower the pH to inhibit the growth of the organisms on the sand?" From these initial tests, I would say that the pH of the bioreactor would have to be lowered to a pH below 4.5 to produce inhibition.

### Mineral Base Medium Results

The microorganisms that were used in this test are

listed in Table 20. Turbidity was used as a means of determining growth. The samples were examined at each day for obvious growth while they were in the shaker. The appearance of obvious growth, approximately  $10^7$  bacteria/ml, took approximately ten days. Table 20 also indicated which microorganisms showed obvious growth on the day they were taken from the shaker (day 11). These bacteria were then used for the mineral base medium plus the hydrocarbon and biocide test.

These results were evaluated each day for obvious growth and after ten days then was no apparent growth. After three weeks in the shaker, none of the samples showed apparent growth, so serial dilution were performed on PCA to see if the bacteria were surviving. The results of the bacterial load in this medium are given in Table 21.

TABLE 20: Bacteria used in the Mineral Base Medium Test

Biotype	Location	Name	Growth in Medium
000004401	B	Aci.lwoffi	yes
sand isolate	G	?	yes
sand isolate	G	?	yes
sand isolate	G	?	yes
sand isolate	G	?	yes
no growth	A	?	yes
000004401	D	Aci.lwoffi	yes
000004012	A	Pseud.spp.	yes
000000020	I	Moraxella	yes
no growth	A	?	yes
no growth	A	resembled Aci.	no
no growth	H	resembled Aci.	no
000000000	F	Aci.lwoffi	no
000000000	F	Aci.lwoffi	no
000000000	E	Aci.lwoffi	no
000000000	E	Aci.lwoffi	no
000000000	M	Aci.lwoffi	no

Note: Sand isolates refer to the ones that were isolated from the pH testing on the sand of the bioreactor. The pH was lowered to 4.5.

**Table 21: Results of Mineral Base Medium and Biocide Bacterial Count**

Biotype	Location	Growth
000004401	B	> 1000 CFU/ml
sand isolate	G	> 1000 CFU/ml
sand isolate	G	> 1000 CFU/ml
sand isolate	G	> 1000 CFU/ml
sand isolate	G	> 1000 CFU/ml
no growth	A	> 1000 CFU/ml
000000401	D	> 1000 CFU/ml
000004012	A	> 1000 CFU/ml
000000020	I	> 1000 CFU/ml
no growth	A	> 1000 CFU/ml

NOTE: When the samples were plated, the dilutions plated were from 10 to the 3rd to 10 to the 6th. The amount of growth on the 10 to the 3rd plates were less than 30 colonies and could not be counted. Thus the number of bacteria in the medium is given as less than a 1000 to indicate that there was survival but no growth. Since 0.1 ml of a 24 hour culture was put into 50 ml of the medium (10 to the 5th), the results indicates a decline in the microbial population. The reasons for the decline might be because the environment was too stressful for survival.

## CONCLUSION

The results obtained from this project revealed many things about the microbial population in the MWF and the waste treatment systems. As expected the microbial load in the bioreactors is much higher than any place else in the system. The control of bacteria in the MWF by biocides is done to prevent the biodeterioration of the MWF and to prevent other contamination problems caused by the microorganisms (table 2). The level in the bioreactors is expected to be higher because it is in the bioreactors that the bacteria are responsible for the biodegradation of industrial waste water.

The bacteria isolated and identified from the samples showed to be Acinetobacter species and Pseudomonas species.

Pseudomonads are one of the most group of important organisms found in MWF [4]. The PASCO ID Tri-Panels were used to identify the microorganisms, however it was not always possible match an organisms' name to the biotype number.

The results from the tests performed on Acinetobacter lwoffii showed that this organisms displayed only slight differences in the pH and the biocide tests. The ratio of the MIC/recommended efficacy level was approximately the same for the three biocides. The test in the mineral base medium showed that Acinetobacter lwoffii isolated from Location B (MWF from the south side) was able to grow in

this medium whereas none of the other Acinetobacter lwoffii tested showed this growth in the 11 days.

The pH tests showed that most of the organisms isolated grew in a pH range of 6 to 8 while some grew at pH 9. The pH tests dealing with bioreactor sand showed that if the microorganisms were left in natural environment, they could survive in an environment with a much lower pH. One microorganism isolated from the sand at pH 4.5 was to able to grow at pH 4.5. This microorganism was not identified at the time.

The results of the biocide tests on other bacteria demonstrated the variance that one would have liked to see with Acinetobacter lwoffii. It must be remembered that these bacteria were all different and from different locations. The data from these bacteria were not as complete as the ones obtained from the Acinetobacter lwoffii thus no conclusion could be drawn. If they had been isolated each location than one would have had data to compare the results. Based on the MIC/recommended efficacy dose, one could say that the bacteria in Table 17 can tolerate a higher level of "Biocide B" than "Biocide C" as indicated in Tables 18 and 19. Thus indicating that Biocide C might be a stronger inhibitor than Biocide B.

The mineral base medium test indicated that some microorganisms could grow in this medium while other could not. The growth of the bacteria in this medium was not depend upon the location the it was isolated from. The

results indicated that it was based upon the type of microorganisms. After three weeks in the mineral base medium and "Biocide B", there was no apparent growth in any of the flasks. To determine if the bacteria survived, plates counts were done on them (Table 21). The results from these counts showed that the bacteria had a harder time surviving in this environment than the two environment separately. This environment might be to stressful for the organisms, even though they did survive in low numbers, to reproduce.

## Appendix 1

### PASCO MIC PANELS, MIC/ID PANELS, BREAKPOINT/ID PANEL and ID TRI-PANEL for use in ANTIMICROBIAL SUSCEPTIBILITY TESTING and IDENTIFICATION of GRAM-NEGATIVE MICROORGANISMS

#### INTENDED USE

PASCO MIC PANELS are used to quantitatively measure the *in vitro* susceptibility of microorganisms to a battery of antimicrobial agents. The principle of the classical broth dilution method is utilized to determine the minimum inhibitory concentration (MIC) of each antimicrobial agent that will inhibit visible growth of an infectious organism *in vitro* 3 to 22 hours. The procedures described herein are based on current recommendations of the National Committee for Clinical Laboratory Standards (NCCLS).<sup>1,2</sup>

The MIC GRAM-NEGATIVE PANEL contains antimicrobial concentrations used primarily to test gram-negative organisms, whereas those contained in the MIC GRAM-POSITIVE PANEL are generally used to test gram-positive organisms. The RESISTANT 5 PANEL is used to test either gram-positive or gram-negative organisms with antimicrobial agents of specific interest.

The PASCO Identification System utilizes an optimized test set of 30 biochemical substrates and normalized thresholds as the statistical basis for identification of the family *Enterobacteriaceae* and other aerobic gram-negative bacilli.<sup>3,4,5,6,7,8,9,10,11,12,13,14,15,16,17,18,19,20,21,22,23,24,25,26,27,28,29,30,31,32,33,34,35,36,37,38,39,40,41,42,43,44,45,46,47,48,49,50,51,52,53,54,55,56,57,58,59,60,61,62,63,64,65,66,67,68,69,70,71,72,73,74,75,76,77,78,79,80,81,82,83,84,85,86,87,88,89,90,91,92,93,94,95,96,97,98,99,100</sup> Three sets of biochemical substrates are used in the ID TRI-PANEL to allow identification of three organisms simultaneously, whereas the PASCO MIC/ID GRAM-NEGATIVE PANEL and BREAKPOINT/ID GRAM-NEGATIVE PANEL contain one set of biochemical substrates along with antimicrobial agents to identify and determine the antimicrobial susceptibility of gram-negative bacilli. Although the BREAKPOINT/ID GRAM-NEGATIVE PANEL fully dilutes ampicillin, gentamicin and tobramycin to facilitate lower dosing of these drugs and thereby decrease the risk to the patient of nephrotoxicity and ototoxicity due to high serum levels, the majority of the antimicrobial agents are included in an abbreviated format of two and three concentrations to allow testing of additional antimicrobials. It has been indicated by NCCLS that, upon using the abbreviated format panels, it is advisable to report a qualitative result (i.e., susceptible, moderately susceptible or resistant). However, the MIC may also be reported if desired.<sup>1,2</sup>

#### PRINCIPLES OF THE PROCEDURE

The MIC test panels contain antimicrobial agents diluted to appropriate test concentrations for blood and soft tissue infections, as well as those agents specifically indicated for lower urinary tract infections. As recommended by NCCLS,<sup>1,2</sup> the diluent for the antimicrobial solutions is Mueller Hinton broth supplemented with calcium and magnesium to ionized physiologic levels. Also as recommended, sodium chloride is added to the broth diluent used for the oxacillin concentrations on the Gram-Positive Panel. This hypertonic broth diluent enhances detection of most oxacillin-resistant or methicillin-resistant staphylococci (MRSA).<sup>25</sup>

The biochemical substrates and reagents used in the PASCO Identification System are based on conventional media formulations.<sup>1,2,7,29,32</sup> Twenty-seven of the 30 tests employed are used to identify oxidase-negative organisms and a second 27-test subset is used to identify oxidase-positive organisms. Specific performance principles for each substrate are provided below.

**CARBOHYDRATE UTILIZATION:** Utilization of the specific carbohydrate results in acid formation. The subsequent pH drop is detected by bromothymol blue indicator which changes from green to yellow.

**INDOLE:** Indole produced from the breakdown of tryptophan by tryptophanase reacts with para-methylaminobenzoic aldehyde (Kovacs' Reagent) to produce a pink to red-colored complex.

**TRYPTOPHAN DEAMINASE:** Indole pyruvic acid, formed by the oxidative deamination of tryptophan, produces a reddish-brown color in the presence of ferric ions.

**ESCULIN:** Hydrolysis of esculin results in the formation of esculetin and glucose. Esculetin combines with ferric ions to produce a brown-colored complex.

**VOGES-PROSKAUER:** Acetoin, produced from sodium pyruvate, combines with creatine and alpha-naphthol at an alkaline pH to produce a red color.

**ARGININE, LYSINE, ORNITHINE:** Anaerobic catabolism of arginine, lysine and ornithine results in formation of the corresponding basic amine, which is detected by brom cresol purple indicator.

**UREA:** Hydrolysis of urea by the enzyme urease results in formation of ammonia. The subsequent pH rise is detected by brom thymol blue indicator which changes from green to blue.

**GLUCOSE FERMENTATION:** Fermentation of glucose results in production of acidic end products. The subsequent pH drop is detected by brom thymol blue indicator which changes from green to yellow.

**ONPG:** Hydrolysis of ortho-nitrophenyl-β-galactopyranoside (ONPG) by beta galactosidase releases yellow ortho-nitrophenol from the colorless ONPG.

**CETRIMIDE:** Resistance to cetrimide is demonstrated by growth in broth supplemented with cetyltrimethyl ammonium bromide.

**MALONATE CITRATE:** Utilization of citrate or malonate as a sole carbon source results in the formation of alkaline end products which change from brom thymol blue indicator from green to blue.

**NITRATE:** Reduction of nitrate to nitrite is detected by the addition of N, N-dimethyl-1-naphthylamine and sulfanilic acid, which combine and react to form a red-colored compound. Complete reduction of nitrate to ammonia or nitrogen gas may be observed in some gram-negative reactions and is confirmed by the addition of zinc dust which detects the presence of unreduced nitrate.

**CEPHALOTHIN, COLISTIN, KANAMYCIN, PENICILLIN, TOBRAMYCIN:** Resistance to specific concentrations of these antimicrobial agents is demonstrated by growth in Mueller Hinton broth containing these agents.

The panels are frozen immediately after preparation and are shipped to the customer in the convenient frozen microtiter format. They must be stored at 20°C or colder in a noncycling freezer. Test panels are thawed prior to use. Inoculated panels may be incubated 16 to 20 hours and observed for visible growth or color changes, as described above, to determine the identification of the organism. The lowest concentration of each antimicrobial agent with no apparent visible growth of the test organism is recorded as the minimum inhibitory concentration (MIC) expressed in micrograms per milliliter (μg/ml).

#### REAGENTS

Refer to Table 3 for current information regarding the contents of each individual panel.

#### PRECAUTIONS

Panels are For *In Vitro* Diagnostic Use

Observe aseptic techniques and established precautions against microbiologic hazards throughout all procedures. Since inoculated panels may contain potentially pathogenic organisms, all materials should be autoclaved prior to disposal.

Precautions for each of the substrate reagents are provided on the respective reagent label.

#### STORAGE

Panels are delivered frozen and must be stored at 20°C or colder in a noncycling freezer. 70°C is specified for certain antimicrobial agents (See Table 3). Once thawed, a panel must be used within 2 hours or discarded. Repeated freezing and thawing may reduce antimicrobial potency.

#### PRODUCT DETERIORATION

Do not use panels that show signs of contamination, thawing during shipment, or that fail to provide proper reactions or endpoints with the quality assurance organisms discussed below. When stored as recommended, the antimicrobials and biochemical substrates will retain their potency until the stated expiration date.

#### SPECIMEN COLLECTION AND PREPARATION

Collect specimens and place on primary isolation media according to normal laboratory practice.<sup>1,2,14</sup> When using the Gram-Negative Identification System, confirm that the isolate is gram-negative and determine its oxidase reaction before proceeding.

#### PROCEDURE

Materials Provided  
Frozen panels

#### Materials Required but not Provided

Brain Heart Infusion Broth, 0.5 ml  
Diluent with Taper, 80, 12.5 ml tube  
MIC Brood Suspension, 12.5 ml tube  
Faded Microscopic Slides, 12.5 ml tube  
Plated culture media  
Mineral oil, heavy  
Oxidase reagent (Kovacs)  
Indole reagent (Kovacs)  
10-12% ferric chloride-TDA Test  
40% potassium hydroxide (VP Test)  
5% alpha naphthol (VP Test)  
0.8% sulfanilic acid (Nitrate Test)  
0.8% N, N-dimethyl-1-naphthylamine (Nitrate Test)  
Zinc dust (Nitrate Test)  
Electric Code  
Pasco Biotryp Codebook  
Worksheet to record patient results  
MIC/ID Quality Assurance Worksheet  
to record assurance organism results  
Inoculating loops, 0.001 ml  
Freezer, 20°C or lower (noncycling)  
Freezer, 70°C when specified  
Incubation strips  
Incubator, 35 ± 1°C (humidified, non-CO<sub>2</sub>)

#### Pasco View Box

Pasco Cover Trays (for use as lids)  
Pasco Disposable Inoculator Set:  
for MIC and MIC/ID panels  
Pasco Disposable Inoculator Set:  
for ID TRI-PANELS  
Vortex mixer  
Bacteriometer or Bunsen burner  
25 μl pipette or pipettor with sterile tips  
Spectrophotometer  
(for turbidity standard technique)  
McFarland standard #0.5  
(for turbidity standard technique)  
Patient MIC Report Form  
Quality assurance organisms:  
*Acinetobacter baumannii* ATCC® 19002  
*Enterobacter cloacae* ATCC® 43091  
*Escherichia coli* ATCC® 25922  
*Escherichia coli* ATCC® 35718  
*Staphylococcus aureus* ATCC® 43068  
*Proteus mirabilis* ATCC® 7022  
*Pseudomonas aeruginosa* ATCC® 27853  
*Pseudomonas aeruginosa* ATCC® 43068  
*Staphylococcus aureus* ATCC® 29213  
*Streptococcus faecalis* ATCC® 29212

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#### PERFORMING MIC AND ID TESTING

**Thawing trays:** Remove needed panels from freezer storage. Remove excess air from the bag of remaining panels, promptly reseal and return to the freezer. Place a clean, empty, cover tray on top of each panel and allow the panels to equilibrate to room temperature before inoculating (approximately 30 minutes). Do not allow the panels to remain at room temperature longer than 2 hours before inoculation.

**Inoculum preparation:** The recommended procedures for inoculum standardization are the Stationary Phase Technique and the Turbidity Standard Technique.

##### Stationary Phase Technique

The bacterial suspension is allowed to grow to its stationary phase, followed by a simple one-step dilution to the desired concentration, as described by Barry et al.<sup>14</sup> Select growth from 5 to 10 isolated colonies on a primary isolation agar plate and prepare a slightly turbid suspension in 0.5 ml of brain heart infusion (BHI) broth and incubate at 35 ± 1°C for 4-6 hours. This will result in approximately 10<sup>8</sup> CFU/ml. Pipet 25 μl of this stationary phase suspension into a screw capped tube containing 12.5 ml Diluent #1 between 80. Upon final dilution in the panel, the bacterial suspension will be approximately 10<sup>7</sup> CFU/ml.

##### Turbidity Standard Technique

Although the inoculum for most organisms can be consistently standardized by the stationary phase method, a more standardized inoculum can be achieved for slower-growing organisms by adjusting the turbidity of a suspension of the organism to that of a McFarland standard #0.5.<sup>1,2</sup> The suspension can be prepared from a broth culture, however, current NCCLS recommendations state that resistant staphylococci can be better detected when the inoculum is prepared from an agar plate.<sup>14</sup> This direct inoculum technique allows detection of a more representative number of the slower-growing oxacillin-resistant strains that are present in small numbers in a heterogeneous culture along with oxacillin-susceptible strains. Perform this procedure as follows:

Prepare the inoculum from growth of a 24-hour culture on agar. Transfer organisms from the agar plate to BHI broth to yield a turbidity equivalent to a McFarland standard #0.5 when using spectrophotometry, nephelometry or visual comparison. This is approximately 10<sup>8</sup> CFU/ml. Pipet 0.25 ml of the standardized suspension into a screw-capped tube containing 12.5 ml Diluent #1 between 80. Invert 8 to 10 times. Upon final dilution in the panel, the bacterial suspension will be approximately 10<sup>7</sup> CFU/ml. (The Promet Inoculator System manufactured by Minnesota Mining and Manufacturing Co. (3M) has been found to be relatively reliable in determining the direct inoculum results.<sup>14</sup> Consult the 3M package insert for further information.)

##### Pasteurized Organisms

NCCLS currently recommends the Turbidity Standard Technique, described above, for preparation of the inoculum of fastidious, slower-growing organisms. The subsequent dilution of the organism suspension should be prepared in 12.5 ml of the appropriate supplemental diluent: a Haemophilus/Fildes-WAGH and PABA (which supplies X and V growth factors) and streptococci, MIC blood supplement (lysed horse blood). Upon inoculation, described below, do not incubate in CO<sub>2</sub> unless necessary for growth. For an explanation of supplements for other fastidious organisms, refer to the NCCLS document.<sup>1,2</sup>

**Tray inoculation:** After the proper dilution is prepared, the bacterial suspension is dispensed into the inoculum tray.

1. Orient the panel so that the labeling may be read, placing the color-coded sterility wells toward the user.
2. Position the inoculator so that the mixing pin will be at the lower right. This orientation assures that the negative control well will not be inoculated.
3. Remove the inoculator from the inoculum tray and gently pour the well mixed bacterial suspension into the inoculum tray, providing for dispersal throughout the channels of the tray. It may be necessary to tilt the tray slightly from front to back to ensure dispersal.
4. Gently lower the inoculator tips into the channels of the inoculum tray and allow all tips to hit by capillary action. It is not necessary to apply pressure to any part of the inoculator.
5. Lift the inoculator, taking care not to touch the tips and lower it with the microbubbles of the panel. When inoculating the ID TRI-PANEL, take care to place the inoculator with the sections delimited on the panel. Orientation of the inoculator is of prime importance to obtain accurate results and the steps must be repeated. Proper alignment at this step is most easily accomplished by holding the inoculator at the corners with the thumb and forefinger of each hand. The 5 μl inoculum will transfer to the wells by capillary action. The 1:20 final dilution of the inoculum suspension results in a microtiter concentration of approximately 10<sup>7</sup> CFU/ml.
6. Return the inoculator to the inoculum tray and dispose of both, along with other contaminated materials.

**Purity and colony count subculturing:** Using a 0.001 ml (1 μl) loop, mix the contents of the positive growth control well and subculture from a onto a suitable plated medium. The resulting colony count should be approximately 100 colonies. If a mixed culture is detected, the MIC and ID test results are invalid.

**Biochemical overlays:** Using a dropper bottle overlay the ONPG, ARG, LYS, ORN and URE wells with 2 drops of heavy weight mineral oil. These wells are underlined on the panel.

**Incubation strip:** Place an incubation strip over the biochemical portion of the panel, taking care not to cover the adjacent antimicrobial wells.

**Incubation:** Place a clean, empty cover tray on each inoculated panel and incubate under humidified conditions (70-80% RH) at 35 ± 1°C for 16-20 hours in a non-CO<sub>2</sub> incubator (unless CO<sub>2</sub> is required for the growth of a fastidious organism). If it is necessary to stack the panels, do so on a noncyclic of five or less to ensure even thermal distribution. Avoid disturbing the placement of the inoculation strip on each panel when adding a cover tray or during stacking. After removing panels from the incubator, either allow them to equilibrate to room temperature before reading or take care to remove condensation from the bottom surface of the plastic tray that might interfere with reading.

#### RESULTS

##### MIC INTERPRETATION

To determine the minimum inhibitory concentration (MIC), read the test panel against an indirectly lighted background. Due to the V shape of the well, bacterial growth in the antimicrobial solutions will always be observed to form a concave white button. However, some bacterial growth (particularly *S. pneumoniae*) demonstrates turbidity throughout the entire circular solution that should be interpreted as growth. Observe the positive growth control to determine the growth typical of the test organism.



- Examine the positive and negative growth controls. The negative growth control well shows no evidence of growth and the positive growth control well shows growth. The end points for the antimicrobial are read.
- No growth in the antimicrobial solution is observed when there is no turbidity or a white precipitate in the well.
- The MIC is recorded as the lowest concentration of antimicrobial solution showing evidence of growth in the growth wells. When reading from the least concentrated dilution to the most concentrated, the MIC is the lowest concentration at which no growth appears. If all dilutions for an antimicrobial demonstrate growth, the end point is recorded as greater than the highest concentration tested. If all dilutions for an antimicrobial show no evidence of growth, the end point for that antimicrobial is recorded as less than or equal to the lowest concentration tested.
- If there is a clear well in a series of growth wells (e.g., growth at 2 and 8 mcg/ml but not at 4 mcg/ml), the MIC should be ignored and the MIC is the next higher mcg/ml (See Figure 1 F). If spot contamination is suspected, as indicated by growth in the negative growth control well or by a pattern of isolated growth wells, the test should be repeated for all antimicrobials.
- The antimicrobial action of such drugs as trimethoprim and sulfas permits growth of a few generations of the organism before inhibition is complete. As a result, a small button of growth may be observed in which case the end point may be interpreted to be the concentration in which there is an 80% reduction in growth when compared with the positive growth control well.

FIG. 1. Illustration of various test results that may occur and the interpretation of each result.

Confirmation of Positive and Negative Growth		Types of Growth	
	Apparent growth in Positive Growth well		Butt
	No growth. Negative Growth well is clear		Diffuse
			Radiated

Higher Concentrations						ROW	EXPLANATION
1	2	3	4	5	6		
A						32	2
B						16	3
C						8	4
D						4	5
E						2	6
F						1	7
G						0.5	8

Lower Concentrations						ROW	EXPLANATION
1	2	3	4	5	6		
A						32	2
B						16	3
C						8	4
D						4	5
E						2	6
F						1	7
G						0.5	8

\* With careful technique, these occurrences are uncommon

\* With careful technique, these observations are uncommon.

## BIOCHEMICAL IDENTIFICATION

Read the biochemical reactions against the white background of the Pasco View Box with the exception of Catalase and the antimicrobial wells, which are read against a black background. Since Cephaloridine, Rifampin, and Rifampicin are used in the MIC portion of the MIC/D panel, but are used for identification purposes, they have been highlighted to facilitate ease of recognition and reading. Examine each biochemical substrate for color change or for the presence of growth and then proceed as follows:

## Criteria for Adding Reagents

- If the isolate is Glucose (GLU) positive after the 16-24 hour incubation, add reagents and read reaction as indicated below.
- If the isolate is Glucose (GLU) negative after the 16-24 hour incubation, but 3 or more alternate tests including Catalase (CAT), Penicillinase (PEN), Cephaloridine (CP) and Tetracycline (TET) are positive, add reagents and read reactions as indicated below.
- If the isolate is Glucose (GLU) negative after the 16-24 hour incubation and less than 3 alternate tests are positive, replace the incubation fluid and incubate for an additional 24 hours before adding reagents.

## Adding Reagents

- To the Vogel-Proskauer (VP) well, add 1 drop of 4% potassium hydroxide followed by 1 drop of 5% alpha-naphthol. Wait 10 minutes before considering the reaction negative.
- To the indole (IND) well, add 2 drops of Kovac's indole reagent. Read within 2 minutes after adding the reagent.
- To the Hydrophobicity (HYP) well, add 1 drop of 10-12% formaldehyde reagent. Read within 2 minutes after adding the reagent.
- To the nitrate (NIT) well, add 1 drop of 0.5% sulfanilic acid and 2 drops of 0.5% N-1-naphthylamine. Read within 2 minutes after adding the reagents. Confirm negative results by adding zinc dust after 10 minutes before considering the reaction positive.

## Organism Identification

Identification of gram-negative organisms may be achieved by using the Pasco Biochemical Identification Chart or by using the use of the Pasco Biotype Codebook or Electric Code™, and which are a derivative of the Pasco Biotype Codebook. The biochemical chart, other than supplemental tests, and all biochemical tests were generated using Pasco's own data base. The Pasco computer-assisted identification system employs 27 oxidative-negative and 27 oxidative-positive biochemical tests. These tests are plotted on three test substrates shown below, which are then used to generate a nine-digit code number for both oxidative-negative and oxidative-positive gram-negative organisms.

## OXIDASE-NEGATIVE TEST SET

OFG	ARA	SOR	RAF	ARG	URE	CT	IND	CET	VALUE
GLU	TRE	VEL	RAA	LYS	CT	ESC	TD	CL	4
MAN	CEL	SUC	ADO	ORN	VAL	OPG	VP	P	1

## OXIDASE-POSITIVE TEST SET

OFG	ARA	SOR	ARC	URE	NIT	IND	CET	K	VALUE
GLU	TRE	VEL	LYS	CT	ESC	TD	CL	CP	4
MAN	CEL	SUC	ORN	VAL	OPG	VP	P	TO	1

The biotype code number is generated by recording the positive and negative reactions and then adding the positive values in each three-test subset. Results obtained in supplemental tests are not used in the biotype number but may be required to confirm identification. A computer-assisted identification service is available at 800-321-9873 for those organisms whose profile numbers are not found in the Pasco Biotype Codebook.

## GUIDELINES TO THE USE OF MIC

- The minimum inhibitory concentration (MIC) reported in programs must be (µg/ml) at the

## Interpretive Criteria for Biochemical Reactions

CODE	TEST	REAGENTS	POSITIVE REACTION	NEGATIVE REACTION	COMMENTS
GLU	Glucose fermentation	None	Yellow to yellow-green	Blue-green or green-yellow	
			42 hr bright yellow	Blue-green or green-yellow	
GLU	Glucose fermentation	None	Yellow to yellow-green	Blue-green or green-yellow	For suspected Pasteurella 40 see below**
MAN	Mannitol	None	Yellow to yellow-green	Blue-green or green-yellow	
ARA	Arabinose	None	Yellow to yellow-green	Blue-green or green-yellow	
TRE	Trehalose	None	Yellow to yellow-green	Blue-green or green-yellow	
CEL	Cellobiose	None	Yellow to yellow-green	Blue-green or green-yellow	
SJA	Sorbitol	None	Yellow to yellow-green	Blue-green or green-yellow	
MEL	Melibiose	None	Yellow to yellow-green	Blue-green or green-yellow	
SUC	Sucrose	None	Yellow to yellow-green	Blue-green or green-yellow	
RAF	Raffinose	None	Yellow to yellow-green	Blue-green or green-yellow	
RAA	Rhamnose	None	Yellow to yellow-green	Blue-green or green-yellow	
ADO	Adonitol	None	Yellow to yellow-green	Blue-green or green-yellow	
ARG	Arginine	None	Fermenters: Purple	Non-fermenters: Purple (see comments)	For non-fermenters, examine the Arg, Lys and Orn wells for any differences in growth.
LYS	Lysine	None	Fermenters: Purple or light to dark gray	Non-fermenters: Purple (see comments)	Apply the reaction to 2 to be significant, and 3 to be significant. The reactions obtained with other amino acids. Negative reactions tend to be yellow to light purple and comparable to each other in color.
ORN	Ornithine	None	Fermenters: Purple or light to dark gray	Non-fermenters: Purple (see comments)	
URE	Urea	None	Blue to blue-green	Yellow, green or green-blue	
CT	Catalase	None	Blue to blue-green	Yellow, green or green-blue	
MAI	Maltase	None	Blue to blue-green	Yellow, green or green-blue	
NIT	Nitrate	2 drops of 0.5% sulfanilic acid and 2 drops of 0.5% N-1-naphthylamine	Pink to red	Colorless	Read within 2 min after addition of reagents. Confirm negative result by adding zinc dust.
		After zinc dust	Colorless	Red	Wait 10 min before considering result.
ESC	Esculin	None	Light to dark brown	Yellow or beige	
GNP	α-glucosidase	None	Yellow	Colorless	Any evidence of growth is positive. The Catalase well should be used as the negative control.
IND	Indole	2 drops of Kovac's indole reagent	Pink to red ring	Colorless to yellow	Read within 2 min after addition of reagent.
TD	Tryptophan deaminase	1 drop of 10-12% formaldehyde	Brown red or orange	Yellow to yellow-orange	Read within 2 min after addition of reagent.
VP	Voges-Proskauer	1 drop of 40% KOH followed by 1 drop of 5% alpha-naphthol	Pink to red	Colorless to yellow	Wait 10 min before considering the reaction negative. Development of a pink color after 10 min may occur with some organisms and should be considered a negative reaction.
CET	Cephaloridine	None	Growth	No growth	Any evidence of growth is compared to the sterility well as positive.
CL	Catalase	None	Growth	No growth	Any evidence of growth is compared to the sterility well as positive.
P	Penicillinase	None	Growth	No growth	Any evidence of growth is compared to the sterility well as positive.
K	Kanamycin	None	Growth	No growth	Any evidence of growth is compared to the sterility well as positive.
CP	Cephaloridine	None	Growth	No growth	Any evidence of growth is compared to the sterility well as positive.
TO	Tetracycline	None	Growth	No growth	Any evidence of growth is compared to the sterility well as positive.

\*\* For an unhydrolyzed description, the color based test is the predominant color. e.g., yellow-green is the color, not green.

\*\* If the isolate demonstrates negative reactions with most of the carbohydrate and the glucose fermentation (GLU) is positive, Pasteurella should be suspected. Under these conditions, carbohydrate reactions should be limited to positive on the basis of any evidence of yellow for this genus only. All other reactions should be read as described above.



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## ABSTRACT

### CHARACTERIZATION OF BACTERIA FOUND IN METAL-WORKING FLUIDS AND THE WASTE TREATMENT SYSTEM INVOLVED IN DEGRADATION OF WASTE WATER

by

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May, 1991

Advisor: Harold W. Rosamoore

Major: Biological Sciences

Degree: Masters of Science

This paper contains information concerning the microbial populations in Metal-working Fluids and the waste treatment system that degrades the waste water. The microorganisms were isolated and identified and then various chemical tests were performed on them to determine the effects it had the microorganisms. These tests included changing the pH of the environment, determining the biocide MIC, and changing the carbon source in the environment. These results were then analyzed to see if microorganisms of the same species but from different location showed any resistant to the tests. Based on the results obtained on Acinetobacter lwoffii, the location of this bacterium did not influence the results.

### AUTOBIOGRAPHIC STATEMENT

Currently, I am a Chemical Officer (Captain) in the United States Army. I was commissioned into the Army on May 1, 1985. I was selected to attend Wayne State University for advanced education in my career field. This selection is based on previous performance in all military assignments, undergraduate degree and grades. I obtained Bachelor of Science from Northern Michigan University, where I studied Biochemistry.

My assignments in the military are outlined below. I attended the chemical officer basic course from June 1985 to October 1985 at Fort McClellan, Al. From here I did a three year tour in Stuttgart, Germany. While in Germany, I was assigned as a platoon leader to the 11th Chemical Company, an assistant chemical officer to the 2d Support Command, and as the officer in charge of the 242d Chemical Detachment. I was awarded the Army Commendation Medal when I departed Germany. My next duty station was back at Fort McClellan for the officer advance course, where I was notified that I was selected to attend graduate school under the fully funded advanced civilian schooling program. Upon completion of my studies at WSU, I will go back to the Army to put this training to use.